B. Tech III Year I Semester

JNTUA COLLEGE OF ENGINEERING (AUTONOMOUS) PULIVENDULA 19ABS41-CHEMISTRY OF ENERGY MATERIALS

(Open Elective-I)

L T P C 3 0 0 3

Course Objectives:

- To make the student understand basic electrochemical principles such as standard electrode potentials, EMF and applications of electrochemical principles in the design of batteries.
- To understand the basic concepts of processing and limitations of fossil fuels and Fuel cells & their applications.
- To impart knowledge to the students about fundamental concepts of hydrogen storage in different materials and liquefaction method
- Necessasity of harnessing alternate energy resources such as solar energy and its basic concepts.
- To understand and apply the basics of calculations related to material and energy flow in the processes.

UNIT – 1: Electrochemical Systems

9 Hrs

Galvanic cell, standard electrode potential, application of EMF, Electrode mechanism, polarization, Batteries-Lead-acid and Lithium ion batteries.

Learning Outcomes:

At the end of this unit, the student will be able to

Solve the problems based on electrode potential
 Describe the Galvanic Cell
 Differentiate between Lead acid and Lithium ion batteries
 Illustrate the electrical double layer

UNIT - II: : Fuel Cells

Basic design of fuel cell, Fuel cell working principle, Fuel cell efficiency Classification of fuel cells, Polymer electrolyte membrane (PEM) fuel cells, Solid-oxide fuel cells (SOFC), and their applications

Learning Outcomes:

At the end of this unit, the student will be able to

Describe the working Principle of Fuel cell
Explain the efficiency of the fuel cell
Discuss about the Basic design of fuel cells
Classify the fuel cell
L2
L3
L2

UNIT – III: Hydrogen Storage

Hydrogen Storage, Chemical and Physical methods of hydrogen storage, Hydrogen Storage in metal hydrides, metal organic frame works (MOF), Carbon structures (Carbon nano tubes, fullerenes), metal oxide porous structures, hydrogen storage by high pressure methods. Liquefaction method

Learning Outcomes:

At the end of this unit, the student will be able to

Differentiate Chemical and Physical methods of hydrogen storage
 Discuss the metal organic frame work
 Illustrate the carbon and metal oxide porous structures
 Describe the liquification methods

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UNIT – IV: Solar Energy

Solar energy introduction and prospects, photovoltaic (PV) technology, concentrated solar power (CSP), Solar Fuels – Hydrogen: Ammonia& Hydrazine, Solar cells (Si-Te& Cd-Te), advantages and disadvantages.

Learning Outcomes:

At the end of this unit, the student will be able to

•	Apply the photo voltaic technology	L3
•	Demonstrate about solar energy and prospects	L2
•	Illustrate the Solar cells	L2
	Discuss about concentrated solar power	L3

UNIT - V: Photo and Photoelectrochemical Conversions

Photochemical cells and applications of photochemical reactions, specificity of photoelectrochemical cell, advantage of photoelectron catalytic conversions.

Learning Outcomes:

At the end of this unit, the student will be able to

•	Differentiate between Photo and Photo electrochemical Conversions	L2
•	Illustrate the photochemical cells	L2
•	Identify the applications of photochemical reactions	L3'
•	Interpret advantages of photoelectron catalytic conversion	L2

Text Books:

- 1. Bahl and Bahl and Tuli, Essentials of Physical Chemistry, S. Chand Publications, New Delhi, 28th Edition, 2020.
- 2. US Department of Energy (EG&G technical services and corporation), Fuel Cell Hand Book 7th Edition, 2004.

Reference Books:

- 1. Ira N. Levine, Physical chemistry 6th Edition, McGraw Hills Education, New Delhi, 2009.
- 2. Silver and Atkins, Inorganic Chemistry, 7th Edition, Oxford University Press, 2018.
- 3. Michael Hirscher, Hand book of Hydrogen Storage: New materials for future energy, storage, Wiley-VCH Verlag GmbH & Co. KGaA, 2010
- 4. Klaus Jagar et.al., Solar energy fundamental, technology and systems, UIT-Cambridge publishers, 2016

Course Outcomes:

At the end of this Course the student will be able to

•	Understand to perform simultaneous material and energy balances	L1
•	Lists about various electrochemical and energy systems	L1
•	Classify solid, liquid and gaseous fuels	L3
	Analyze the energy demand of world, nation and available resources to fulfill the demand	L3
•	Evaluate the conventional energy resources and their effective utilization	L3
•	To be able to understand and perform the various characterization techniques of fuels	L1
•	Explain knowledge of modern energy conversion technologies	L2
•	To be able to identify available nonconventional (renewable) energy resources and techniques to utilize them effectively	L1

